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Reactions of 4-Methoxy-3-quinolinyl and 1,4-Dihydro-4-oxo-3-quinolinyl Sulfides Aiming at the Synthesis of 4-Chloro-3-Quinolinyl Sulfides [1]

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The preparation of 1,4-dihydro-4-oxo-3'-alkylthio-3,4'-diquinolinyl sulfides 3 or 1,4-dihydro-4-oxo-3-(alkylthio)quinolines 4 by acid catalysed hydrolysis of 4-methoxy-3'-alkylthio-3,4'-diquinolinyl sulfides 1 or 4-methoxy-3'-alkylthio-3,4'-diquinolinyl sulfides 1 or 1,4-dihydro-4-oxo-3'-alkylthio-3,4'-diquinolinyl sulfides 3 with phosphoryl chloride in DMF afforded 4-chloro-3'-alkylthio-3,4'-diquinolinyl sulfides 5. Treatment of the title compounds 1 or 3 with boiling phosphoryl chloride systems:leads to 4-chloro-3-(alkylthio)quinolines 6 and thioquinanthrene but those of alkoxy- or oxo-quinolines 2 or 4 lead to 4-chloro-3-(alkylthio)quinolines 6. The reactions of N-methyl-4(lH)-quinolinones 3n and 4n with phosphoryl chloride directed to 4-chloro-3-(alkylthio)quinolines 6 were studied as well.

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Introduction.

Typical transformations of the substituents being in the aza-activated positions in six-membered hetarenes were usually carried out from x-chloro-systems:to x-alkoxy ones [2-3]. However, our recent results made 4-alkoxy-3quinolinyl sulfides 1 and 2 easily available [4-5]. It induced the present study on the reverse substituent replacement, i.e. from 4-alkoxyquinolines 1 and 2 to 4chloroquinolines 5 or 6, indirectly via 4-quinolinones 3 or 4 and then directly using phosphoryl chloride as a chlorine source. Two routes were elaborated: the first one was performed under mild conditions in DMF and at room temperature. It allowed for the simple direct replacement of the 4-alkoxy or 4-oxo- group for the 4-chloro one, both in quinolinyl sulfides 2 and 4 as well as in 3',4-diquinolinyl bis-sulfides 1 or 3. Under more rigorous conditions, i.e. in the boiling phosphoryl chloride/triethylamine hydrochloride system, the replacement mentioned above took place also but the reaction of compounds 1 and 3 ran with the cleavage of both γ-quinolinyl-heteroatom bonds and led to 4-chloro-3-(alkylthio)quinolines 6 and thioquinanthrene. For a comparison, the reactions of selected N-methyl-4(1H)-quinolinones 3 \mathbf{n} and 4 \mathbf{n} with phosphoryl chloride were performed as well.

Results and Discussion.

The reactions of 4(1H)-quinolinones or quinoline N-oxides are the most often exploited methods for the preparation of 4-chloroquinolines [2,6]. Since 3-alkylthio-4(1H)-quinolinones 3 and 4 are easily available by the acid hydrolysis of 4-alkoxy-3-quinolinyl sulfides 1 [4] or

2 (see Schemes 1 and 2, and experimental), we started with the reactions of compounds 3 and 4 with boiling phosphoryl chloride. Compounds 4 gave pure 4-chloro-3-(alkylthio)quinolines 6 (78-89%). However, similar treatment of diquinolinyl sulfides 3a-c led to 4-chloro-3-(alkylthio)-quinolines 6 (65-80%) and thioquinanthrene (86-90%). Most probably, under acid catalysis conditions, the reactions of compounds 3a-3c run by the cleavage of both γ -quinolinyl-heteroatom bonds to form two 4-chloro-quinolines 6h and 6a-6c. The first one should undergo cyclodehydrochlorination to form the 1,4-dithiin ring of thioquinanthrene as it was suggested or proved in the case of the formation of dithiinodipyridines [7], dithiinodiquinolines [8] and dithiinodipyridazines [9] from respective ortho chloromercaptoazines (see Scheme 2).

Since both *N*-non-substituted lactams [6] and (but more difficult) *N*-alkyl ones [6] can be transformed into the respective chlorolactim derivatives by the reactions with phosphoryl chloride, we attempted to carry out the reactions with *N*-alkyl-4(1*H*)-quinolinones **3n** and **4n**. However, complete consumption of 1-methyl-3-methylthio-4(1*H*)-quinolinone **4n** (directed in 85% in 4-chloro-3-(methylthio)quinoline **6a**) required 20 hours as compared to 0.5 hour for *N*-non-substituted 4-quinolinone **4a**.

Furthermore, the reactions of 1-methyl-1,4-dihydro-4-oxo-3'-methylthio-3,4'-diquinolinyl sulfide 3n with phosphoryl chloride up to 2 hours gave only 4-chloro-3-(methylthio)-quinoline 6a as the sole isolated product (after treatment of reaction mixture with water followed by neutralization). Prolongation of the reaction time up to 72 hours gave both 4-chloro-3-(methylthio)quinoline 6a (80%) and thioquinanthrene (37%). It indicates that the reactions of 3,4'-diquinolinyl sulfide 3n having a N-methyl-4(1H)-quinolinone fragment also took place with the breaking of the γ -quinolinyl sulfur bond to give 4-chloro-3-(methylthio)quinoline 6a as the product formed from the bis-sulfide part of the substrate 3n and a thioquinanthrene as the product formed by N-demethylation of the N-methyl 4(1H)-quinolinone fragment of 3n.

Comparing the results of the conversion of N-non-substituted 4(1H)-quinolinones $4\mathbf{a}$ - \mathbf{c} and N-methyl-4(1H)-quinolinone $4\mathbf{n}$ into 4-chloroquinolines of type 6 versus those in the transformation of the N-non-substituted 4(1H)-quinolinone fragment of $3\mathbf{a}$ - \mathbf{c} and the N-methyl-substituted 4(1H)-quinolinone fragment of $3\mathbf{n}$ into thio-quinanthrene one could conclude that the N-demethylation process seems to be the step limiting further transformation of N-methyl-4(1H)-quinolinones $3\mathbf{n}$ and $4\mathbf{n}$. To explain the unsatisfactory balance of both the quinoline part of $3\mathbf{n}$ as well as the low yield in the formation of thioquinanthrene from $3\mathbf{n}$ in its reaction with phosphoryl

Table
Reactions of 4-Methoxy-3-quinolinyl Sulfides 1 and 2 and 1,4-Dihydro-4-oxo-3-quinolinyl Sulfides 3 and 4 with Phosphoryl Chloride.

Entry	Substrate	Chlorinating system	Temp. [°C]	Time [hour(s)]	The products, yield (%)	
					4-Chloroquinoline	Thioquinanthrene*
1	la R=Me	Λ	reflux	0.5	6a (80)	95
2	la R=Me	В	r.t.	24	5a (58)	
3	1b R=Et	Α	reflux	0.5	6b (80)	95
4	1b R=Et	В	r.t.	24	5b (57)	
5	1c R=PhCH ₂	Α	reflux	0.5	6c (70)	96
6	1c R=PhCII ₂	В	r.t	24	5c (57)	
7	2a R=Me	Α	reflux	0.5	6a (94)	
8	2b R=Et	Α	reflux	0.5	6b (91)	
9	2c R=PhCH ₂	Α	reftux	0.5	èс (87)	
10	3a R=Me	Α	reflux	0.5	6a (80)	86
11	3a R=Me	В	r.t.	24	5a (82)	
12	3b R=Et	Α	reflux	0.5	6b (72)	87
13	3b R=Et	В	r.t.	24	5b (76)	
14	3c R=PhCH ₂	Α	reflux	0.5	6c (65)	90
15	3c R=PhCH,	В	r.t.	24	5c (69)	
16	3n	A or A1	reflux	- 0.5	6a (86)	0
17	3n	A or A1	reflux	5	6a (85)	4
18	3n	A or A1	reflux	30	6a (84)	21
19	3n	A or A1	reflux	72	6a (80)	37
20	4a R=Me	Α	reftux	0.5	6a (89)	
21	4a R=Me	H	r.t.	24	6a (78)	
22	4b R=Et	Λ	reflux	0.5	6b (86)	
23	4c R=PhCH ₂	Α	reflux	0.5	6c (79)	
24	4n	Α	reflux	5	6a (65)	
25	4n	Α	reflux	20	6a (85)	

A - POCl₃(neat), A1 - POCl₃/Et₃N•HCl, B - POCl₃ in DMF. * IUPAC name.

chloride we considered the occurence of 5,12-dimethylthioquinanthrenediinium bis-salts as a percursor of thioquinanthrene. However, treatment of 5,12-dimethylthioquinanthrenediinium bis-chloride with boiling phosphoryl chloride did not lead to thioquinanthrene although it exhibited high instability of the starting bis-salt, which was 50% consumed after 1 hour and completely consumed after 24 hours. In the last case only the formation of traces of thioquinanthrene was noted. (Further study of the behaviour of 5,12-dimethylthioquinanthrenediinium bis-salts in boiling phosphoryl chloride is now in progress).

The lability of the 4-alkoxy group in 4-alkoxyquinolines under acid catalysis conditions [10,4], and more generally, the sensitivity of the aza-activated ether linkage in alkoxy or aryloxyheteroarenes towards acid catalysed nucleophilic displacement [1 l] directed us to the reactions of 4-alkoxyquinolines 1 and 2 with phosphoryl chloride. In fact, treatment of 4-methoxy-3-(alkylthio)quinolines 2a-2c with boiling phosphoryl chloride during 0.5 hour afforded 4-chloro-3-(alkylthio)quinolines 6a-6c in high yields (see Scheme 1). The reaction of 4methoxy-3'-alkylthio-3,4'-diquinolinyl sulfides 1 with boiling phosphoryl chloride gave the same results as in the case of quinolinones 3a-3c, i.e. 4-chloro-3-(alkylthio)quinolines 6 and thioquinanthrene were formed (Scheme 2). Also reactions of 4-alkoxydiquinolinyl sulfides 1a-1c with the phosphoryl chloride/DMF system led to the same 4-chloroquinoline products 5a-5c as for compounds 3, however, with some lower yields (52-58%) (Scheme 2).

Differences in the reactivity of 4-methoxy-3'-alkylthio-3,4'-diquinolinyl sulfides I versus 1-methyl-1,4-dihydro-4-oxo-3'-methylthio-3,4'-diquinolinyl sulfide 3n towards phosphoryl chloride also support our revision [4-5] that 4-alkoxy-3'-alkylthio 3,4'- diquinolinyl sulfides 1 but not 1-alkyl-1,4-dihydro-4-oxo-3'-alkylthio-3,4'-diquinolinyl sulfides 3 [16] are the products of the reaction of thioquinanthrene with sodium alkoxides followed by S-alkylation.

In conclusion, we have shown that 4-chloro-3'-alkylthio-3,4'-diquinolinyl sulfides 5 and(or) 4-chloro-3-(alkylthio)quinolines 6 can be easy obtained by an effective 3 or 4 step synthesis starting from quinoline *via* thio-

quinanthrene.

EXPERIMENTAL

Melting points were determined in open capillary tubes on a Boetius melting point apparatus and are uncorrected. The ¹H nmr spectra were recorded on a Bruker MSL 300 (300 MHz) spectrometer in deuteriochloroform or DMSO-d₆ solvents with tetramethylsilane as the internal standard and chemical shifts reported in ppm (δ) and J values in IIz. EI mass spectra were run on a LKB GC 2091 spectrometer at 70 eV and 15 eV. Thin layer chromatography was performed on silica gel 60 254F plates (Merck) using a mixture of chloroform and ethanol (15:1 v/v) as an eluent.

Thioquinanthrene was obtained by exhaustive sulfurization of quinoline with elemental sulfur [12]. l-Methyl-l,4-dihydro-4oxo-3'-methylthio-3,4'-diquinolinyl sulfide 3n was prepared by methylation of the sodium salt of 1,4-dihydro-4-oxo-3'-methylthio-3,4'-diquinolinyl sulfide 3a according to the method described earlier [4]. l-Methyl-3-(methylthio)-4(1H)-quinolinone 4n with mp 123-124° and ¹H nmr as in ref [4] was prepared by thermal rearrangement of 4-methoxy-3-(methylthio)quinoline 2a [13]. 4-Methoxy-3-(methylthio)quinoline 2a, 4-methoxy-3-(ethylthio)quinoline 2b, 4-methoxy-3-(benzylthio)quinoline 2c, were prepared from thioquinanthrene and sodium methoxide in DMF solution followed by treatment of sodium 4-methoxy-3quinolinethiolate formed with alkylating agents according to the one-pot procedure described previously [5]. 5,12-Dimethylthioquinanthrenediinium bis-chloride was obtained from 5.12-dimethylthioquinanthrenediinium bis(methyl sulfate) [14].

4-Methoxy-3'-alkylthio-3,4'-diquinolinyl Sulfides (1).

Compounds 1 were prepared from thioquinanthrene [12] and sodium methoxide followed by S-alkylation of sodium 4-[(4-methoxy-3-quinolinyl)thio]-3-quinolinethiolate with alkyl iodides (in the case of compound 1c with benzyl chloride) according to the general procedure reported previously [4,5].

4-Methoxy-3'-methylthio-3,4'-diquinolinyl Sulfide (1a).

Compound 1a had mp 131-l32°, yield 91%; ¹H nmr and ms spectra as in ref [4].

4-Methoxy-3'-ethylthio-3,4'-diquinolinyl Sulfide (1b).

This compound had mp 109-1 10° , yield 80%; ¹H nmr (deuteriochloroform): δ 1.34 (t, 3H, J = 7.3 Hz, CH₃CH₂S), 3.1 I (q, 2H, J = 7.3 Hz, CH₃CH₂S), 4.24 (s, 3H, CH₃O), 7.41-8.37 (m, 8H, Ar-H), 8.08 (s, 1H, H-2), 8.81 (s, 1H, H'-2); ms:(70 eV) m/z (relative intensity): 378 (37.8, M⁺), 317 (6.3, M-C₂H₅S), 302 (7.4, M-C₂H₅S-CH₃).

Anal. Calcd. for C₂₁H₁₈N₂OS₂: C, 66.65; H, 4.80; N, 7.41; S, 16.91. Found: C, 66.72; H, 4.87; N, 7.37; S, 16.75.

4-Methoxy-3'-benzylthio-3,4'-diquinolinyl Sulfide (1c).

This compound had mp 99-100°, yield 88%; ${}^{1}H$ nmr (deuteriochloroform): δ 4.13 (s, 3H, CH_3O), 4.27 (s, 2H, CH_2S), 7.17-7.25 (m, 5H, Ar-H), 7.49-8.36 (m, 8H, Ar-H), 8.12 (s, lH, H-2), 8.86 (s, lH, H'-2); ms: (70 eV) m/z (relative intensity) 440 (72.9, M⁺), 409 (18.9, M-CH₃O), 349 (19.4, M-C₆H₅CH₂), 318 (12.1, M-C₆H₅CH₂-CH₃O).

Anal. Calcd. for C₂₆H₂₀N₂OS₂: C, 70.89; H, 4.58; N, 6.36; S, 14.53. Found: C, 70.8; H, 4.47; N, 6.47; S, 14.4.

Hydrolysis of 4-Methoxy-3'-alkylthio-3,4'-diquinolinyl Sulfides 1 to 1,4-Dihydro-4- oxo-3'-alkylthio-3,4'-diquinolinyl Sulfides 3.

A mixture of 4-methoxy-3'-alkylthio-3,4'-diquinolinyl sulfide (1) (3 mmoles) and azeotropic hydrochloric acid (20 ml) was heated at reflux for 15 minutes. The solution was then evaporated *in vacuo* to dryness. The residue was neutralized with 5% aqueous sodium bicarbonate solution (20 m1). The resultant solid was filtered off and air-dried to give crude 1,4-dihydro-4-oxo-3'-alkylthio-3,4'-diquinolinyl sulfide 3, which was crystalized from DMF to yield pure 3.

1,4-Dihydro-4-oxo-3'-methylthio-3,4'-diquinolinyl Sulfide (3a).

Compound 3a had mp 270-272°, ref [4] mp 278-280°, yield 81%; III nmr and ms spectra as in ref [4].

1,4-Dihydro-4-oxo-3'-ethylthio-3,4'-diquinolinyl Sulfide (3h).

This compound had mp 240-242°, yield 79%; ${}^{1}H$ nmr (dimethyl sulfoxide-d₆): δ 1.47 (t, 3H, J = 7.3 Hz, CH₃CH₂S), 3.30 (q, 2H, J = 7.3 Hz, CH₃CH₂S), 7.44-8.54 (m, 8H, Ar-H), 7.77 (s, III, H-2), 8.96 (s, 1H, II'-2); ms: (70 eV) m/z (relative intensity) 364 (43.0, M⁺), 303 (100, M-C₃H₅S).

Anal. Calcd. for C₂₀H₁₆N₂OS₂: C, 65.92; H, 4.43; N, 7.69; S, 17.56. Found: C, 65.98; H, 4.38; N, 7.62; S, 17.67.

1,4-Dihydro-4-oxo-3'-benzylthio-3,4'-diquinolinyl Sulfide (3c).

This compound had mp 130-132°, yield 74%; ^{1}H nmr (dimethyl sulfoxide-d₆): δ 4.57 (s, 2H, CH₂S), 7.25-7.46 (m, 5H, C₆H₅), 7.61-8.57 (m, 8H, Ar-H), 7.79 (s, 1H, H-2), 8.98 (s, 1H, H'-2); ms: (70 eV) m/z (relative intensity) 426 (40.2, M⁺), 335 (12.5, M-C₆H₅CH₂), 303 (100, M-C₆H₅CH₂S).

Anal. Calcd. for C₂₅H₁₈N₂OS₂: C, 70.41; H, 4.26; N, 6.57; S, 15.01. Found: C, 70.58; H, 4.20; N, 6.51; S, 15.20.

Hydrolysis of 4-Methoxy-3-(alkylthio)quinolines 2 into 1,4-Dihydro-4-oxo-3-(alkylthio)quinolines 4.

The same procedure as in the case of transformation of 4-methoxy-3'-alkylthio-3,4'-diquinolinyl sulfides 1 to 1,4-dihydro-4-oxo-3'-alkylthio-3,4'-diquinolinyl sulfides 3 was applied for the conversion of 4-methoxy-3-(alkylthio)quinolines 2 into 1,4-dihydro-4-oxo-3-(alkylthio)quinolines 4.

1.4-Dihydro-4-oxo-3-(methylthio)quinoline (4a).

This compound had mp 170-172°, ref [15] mp 173-175°, yield 93%; 1 H nmr (deuteriochloroform): δ 2.40 (s, 3H, C H_3 S), 7.42-8.53 (m, 4H, Λ r-H), 8.34 (s, 1H, H-2), 12.90 (broad singlet, 1H, NH); ms: (70 eV) m/z (relative intensity) 191 (100, M+), 176 (35.2, M-CH₃), 158 (28.6, M-SH).

1,4-Dihydro-4-oxo-3-(ethylthio)quinoline (4b).

This compound had mp 158-159°, yield 92%; ¹H nmr (deuteriochloroform): δ 1.22 (t, 3H, J = 7.4 Hz, CH₃CH₂S), 2.85 (q, 2H, J = 7.4 Hz, CH₃CH₂S), 7.43-8.45 (m, 4H, Ar-H), 7.97 (d, 1H, J = 8.4 Hz, II-2), 12.96 (broad singlet, 1H, NII); ms: (70 eV) m/z (relative intensity) 205 (100, M+), 191 (21.8, M-CH₂), 176 (37.5, M-C₂H₅).

*Anal.*Calcd. for C₁₁H₁₁NOS: C, 64.36; H, 5.40; N, 6.82; S, 15.62. Found: C, 64.45; H, 5.47; N, 6.71; S, 15.41.

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1,4-Dihydro-4-oxo-3-(benzylthio)quinoline (4c).

This compound had mp 186-188°, yield 89%; ¹H nmr (deuteriochlorofrom): δ 3.87 (s, 2H, $C_6H_5CH_2$), 7.08 (s, 5H, $C_6H_5CH_2$), 7.39-8.47 (m, 5H, Ar-H), 12.42 (broad singlet, 1H, NH); ms:(70 eV) m/z (relative intensity) 267 (25.3, M⁺), 234 (10.4, M-SH), 176 (8.4, M- $C_6H_5CH_2$).

Anal. Calcd. for C₁₆H₁₃NOS: C, 71.88; H, 4.90; N, 5.24; S, 11.99. Found: C, 71.96; H, 4.84; N, 5.18; S, 11.79.

Synthesis of 4-Chloro-3-(alkylthio)quinolines 6 from 4-Methoxy-3-(alkylthio)quinolines 2 or from 1,4-Dihydro-4-oxo-3-(alkylthio)quinolines 4.

A mixture of 4-methoxy-3-(alkylthio)quinoline 2 (10 mmoles) or 1,4-dihydro-4-oxo-3-(alkylthio)quinoline 4 (10 mmoles) and phosphoryl chloride (20 ml) was refluxed for 0.5 hour. Then the excess of phosphoryl chloride was evaporated in vacuo. The residue was carefully poured onto ice (30 g) and then neutralized with conc. ammonia at 0-5°. The solid was filtered off, washed with water and air-dried. The product was purified by vaccum distillation (compound 6a) or crystallized from ethanol (compounds 6b, 6c) to give 86-94% of final product 6.

- 4-Chloro-3-(methylthio)quinoline 6a had b.p. 164-166°/ 1 mm Hg, mp 104-105° (ethanol), ref [16] mp 104-105°.
- 4-Chloro-3-(ethylthio)quinoline **6b** had mp 49-50° (ethanol), ref [16] mp 49-50°.
- 4-Chloro-3-(benzylthio)quinoline 6c had mp 96-97° (ethanol), ref [16] mp 96-97°.

Reaction of 4-Methoxy-3'-alkylthio-3,4'-diquinolinyl Sulfides 1 or 1,4-Dihydro-4-oxo-3'-alkylthio-3,4'-diquinolinyl Sulfides 3 with Boiling Phosphoryl Chloride or with Phosphoryl Chloride/Triethylamine Hydrochloride System.

A mixture of sulfide 1 or 3 (5 mmoles) triethylamine hydrochloride (2.2 g) and phosphoryl chloride (15 ml) was refluxed for 0.5 hour. Then the excess of phosphoryl chloride was evaporated *in vacuo*. The residue was carefully poured onto ice (30 g) and then neutralized with concentrated ammonia at 0-5°. The solid was filtered off, air-dried and boiled with tetrachloromethane (20 ml). The insoluble solid was hot-filtered, air-dried and crystallized from DMI² to give thioquinanthrene (86-96%) with mp 310-311, ref [12] mp 314-315°

The tetrachloromethane filtrate was evaporated to dryness. The residue was crystallized from ethanol to give 4-chloro-3-(alkylthio)quinoline 6 in yields of 65-80%. Results are collected in the Table.

Reaction of 1-Methyl-3-[(3-methylthio-4-quinolinyl)thio]-4-oxo-1,4-dihydro- quinoline 3n with the Phosphoryl Chloride/Friethylamine Hydrochloride System

A mixture of sulfide 3n (1.82 g, 5 mmoles), triethylamine hydrochloride (2.2g) and phosphoryl chloride (20 ml) was refluxed in an oil bath at 140-145° for 0.5-72 hours. Then the excess of phosphoryl chloride was evaporated *in vacuo*. The residue was treated as in the case of the reaction of compound 3a with phosphoryl chloride. Results are collected in the Table.

a) Synthesis of 4-Chloro-3'-alkylthio-3,4'-diquinolinyl Sulfides 5 from 1,4-Dihydro-4-oxo-3'-alkylthio-3,4'-diquinolinyl Sulfides 3.

A mixture of 1,4-dihydro-4-oxo-3'-alkylthio-3,4'-diquinolinyl sulfide 3 (2 mmoles), phosphoryl chloride (0.55 ml) and dimethylformamide (10 ml) was stirred at room temperature under nitrogen atmosphere for 24 hours and poured into a mixture of water and ice (20 ml). The mixture was then neutralized with concentrated aqueous ammonia at 0°. The solid formed was filtered off, washed with water and air-dried. The crude product was then crystallized from dimethylformamide to give the compound 5 in the yields of 69-82%.

4-Chloro-3'-methylthio-3,4'-diquinolinyl Sulfide (5a).

This compound had mp 155-156°, yield 82%; ¹H and ¹³C nmr spectral data were published previously [17]; ms: (15 eV) m/z (relative intensity) 370 (34.2, M + 2), 368 (51.8, M⁺), 335 (13.0, M + 2^{-35} Cl), 333 (100, M-³⁵Cl and M + 2^{-37} Cl); 318 (59.5, M-CH₄Cl).

Anal. Calcd. for $C_{19}H_{13}N_2S_2Cl$: C, 61.95; H, 3.56; N, 7.61; S, 17.38; Cl, 9.50. Found: C, 61.80; H, 3.57; N, 7.57; S, 17.52; Cl, 9.61.

4-Chloro-3'-ethylthio-3,4'-diquinolinyl Sulfide (5b).

This compound had mp 116-117°, yield 76%; ¹H nmr (deuteriochlorofrom): δ 1.37 (t, 3H, J = 7.4 Hz, CH₃CH₂S), 3.1 5 (q, 2H, J = 7.4 Hz, CH₃CH₂S), 7.52-8.32 (m, 8H, Ar-H), 7.96 (s, 1H, H-2), 8.91 (s, 1H, H-2'); ms: (15 eV) m/z (relative intensity) 384 (14.5, M + 2), 382 (34.2, M⁺), 346 (100, M-³⁵Cl and M + 2-³⁷Cl), 318 (91.3, M-Cl-C₂H₅).

Anal. Calcd. for $C_{20}H_{15}N_{2}S_{2}Cl$: C, 62.82; H, 3.96; N, 7.33; S, 16.74; Cl, 9.15. Found: C, 62.75; H, 3.91; N, 7.26; S, 16.85; Cl, 9.22.

4-Chloro-3'-benzylthio-3,4'-diquinolinyl Sulfide (5e).

This compound had mp 106-107°, yield 69%; ¹H nmr (deuteriochloroform): δ 4.31 (s, 2H, CH₂S), 7.20-7.29 (m, 5H, Δ r-H), 7.50-8.29 (m, 8H, Δ r-H), 7.91 (s, 1H, H-2), 8.90 (s, 1H, H-2'); ms: (15 eV) m/z (relative intensity) 446 (6.2, M+2), 444 (15.1, M⁺); 409 (40.6, M-³⁵Cl and M + 2-³⁷Cl), 318 (14.1, M-Cl-C₆H₅CH₂).

Anal. Calcd. for $C_{25}H_{17}N_2S_2Cl$: C, 67.56; H, 3.86; N, 6.31; S, 14.40; Cl, 7.87. Found: C, 67.47; H, 3.87; N, 6.38; S, 14.51; Cl, 7.80.

b) Synthesis of 4-Chloro-3'-alkylthio-3,4'-diquinolinyl Sulfides 5 from 4-Methoxy-3'-alkylthio-3,4'-diquinolinyl Sulfides 1.

The reactions of 4-methoxy-3'-alkylthio-3,4'-diquinolinyl sulfides 1 with phosphoryl chloride in dimethylformamide solution were performed in the same manner as above (procedure a), but in order to obtain pure 5, crude product 5 (upper $R_{\rm p}$ value) should be separated from unreacted methoxy substrate 1 (lower $R_{\rm f}$ value) by column chromatography on silica gel 40 (70-230 mesh), using a mixture of chloroform and ethanol (l0:1 v/v) as an eluent and finally purified by crystallization from DMF. The results in the preparation of 5 are collected in the Table.

REFERENCES AND NOTES

- [1] Part XXVII in the series of Azinyl Sulfides. Part XXVI. K. Pluta, J. Heterocyclic Chem., submitted.
- [2] A. R.Katritzky, Handbook of Heterocyclic Chemistry, Pergamon Press, Oxford, 1985, Chapter 3.2.
- [3] H. Lettau, Chemie der Heterocyclen, VEB Verlag für Grundstoffindustrie, Leipzig, 1980, Chapters 10 and 7.

- [4] S.Boryczka, A. Maślankiewicz, M. Wyszomirski, T. Borowiak and M. Kubicki, *Recl. Trav. Chim. Pays-Bas.* 109, 509 (1990).
- [5] A. Maślankiewicz, and S.Boryczka, Recl. Trav. Chim. Pays-Bas, 112, 519 (1993).
- [6] R. K. Smalley, The Chemistry of Heterocyclic Compounds, Vol.32, Quinolines, Part I, G. Jones, ed, John Wiley and Sons, London, 1977, Chapter 3.
- [7] K. Krowicki and P. Nantka-Namirski, Rocz. Chem., 51, 2435 (1977).
- [8] B. A. Dreikorn, A. F. Elsasser and G. P. Jourdan, J. Org. Chem., 44, 877 (1979).
- K. Kaji and M. Kuzuya, Chem. Pharm. Bull., 18, 970 (1970);
 M. Kuzuya and K. Kaji, Chem. Pharm. Bull., 18, 2420 (1970).

- [10] L. Janda, J. Nguyen, S. E. Patterson, and L. Strekowski, J. Heterocyclic Chem., 29, 1753 (1992).
- [11] R. G. Shepherd and J. L. Fedrick, Adv. Heterocyclic Chem., Vol. IV, A. R. Katritzky, ed, Academic Press, New York, 1965, p 146.
- [12] A. Maślankiewicz, Pol. J. Chem., 59, 511 (1985).
- [13] A. Maślankiewicz, and K. Pluta, Monatsh. Chem., 114, 281 (1983).
 - [14] A. Maślankiewicz and K. Pluta, Heterocycles, 32, 247 (1992).
- [15] D. T. Connor and M. von Strandtmann, J. Heterocyclic Chem., 15, 113 (1978).
 - [16] A. Maślankiewicz and A. Zieba, Synthesis, 872 (1982).
- [17] M. Wyszomirski, A. Gogoll, A. Maślankiewicz and S. Boryczka, *Phosphorus, Sulfur Silicon*, **59**, 225 (1991).